

Europäisches Patentamt

European Patent Office

Office européen des brevets



(11) EP 0 943 649 A2

(12)

EUROPEAN PATENT APPLICATION

(43) Date of publication: 22.09.1999 Bulletin 1999/38

(21) Application number: 99104011.4

(22) Date of filing: 12.03.1999

(51) Int. Cl.⁶: **C08K 5/00**, C08K 5/13, C08K 5/52, C08L 67/02

(84) Designated Contracting States:

AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU

MC NL PT SE

Designated Extension States:

AL LT LV MK RO SI

(30) Priority: 17.03.1998 IT MI980537

(71) Applicant: SINCO RICERCHE S.p.A. / 28048 Verbania Pallanza (Verbania) (IT)

(72) Inventors:

 Al Ghatta, Hussein 03014 Fiuggi (Frosinone) (IT)

Pinto, Antonio
 00042 Anzio (Roma) (IT)

(74) Representative:
 Modiano, Guido, Dr.-Ing. et al
 Modiano & Associati SpA
 Via Meravigli, 16
 20123 Milano (IT)

(54) Stabilized polyester resins having reduced formaldehyde content

(57) Polyester resins with a low formaldehyde content, to which an antioxidant acting as free-radical scavenger is added, containing less than 150 ppb of formaldehyde and having a formaldehyde generation rate of less than 15 ppb/min at 290°C.

Description

5

[0001] The present invention relates to a method for preparing polyester resins with a low content of residual formal-dehyde and a low formaldehyde generation rate, to the resulting resins and to the articles manufactured with said resins.

[0002] Formaldehyde and other contaminants, mainly acetaldehyde, are generated in considerable amounts during resin treatment steps such as blow-extrusion and blow-injection.

[0003] The residual formaldehyde content of the products can exceed 1000 ppb and the formaldehyde generation rate, measured at 290°C, can be higher than 500 ppb/minute.

[0004] The presence of formaldehyde in drinks and/or food, like the presence of other contaminants, negatively affects their organoleptic properties, specifically their taste.

[0005] Accordingly, it is desirable to have the lowest possible formaldehyde content in the resin that makes contact with food or beverages.

[0006] It is also desirable for articles which do not make contact with food or beverages also to have a low residual formaldehyde content.

[0007] In order to obtain articles having a low residual formaldehyde content, the initial resin must be characterized not only by a low aldehyde content but also by a low rate of aldehyde generation at the process temperatures used to prepare the articles.

[0008] Methods are known for obtaining polyester resins with a low acetaldehyde content and with a low acetaldehyde regeneration rate (EP-B-41035).

[0009] The polyester resins meant for blow-extrusion, blow-injection or other processes have intrinsic viscosities usually above 0.75 dl/g. These values are obtained by subjecting the resin, after its polycondensation reaction (in which the intrinsic viscosity generally does not reach values higher than 0.6-0.7 dl/g) to solid-state polycondensation (SSP).

[0010] The SSP reaction is performed by heating the pellets of the resin, after its crystallization, to a temperature from about 180° to 230°C for a time sufficient to obtain the intended IV value.

[0011] The process is performed in a stream of inert gas (nitrogen or other gases).

[0012] Extrusion of the resin to obtain the pellets to be subjected to SSP is generally performed in air. Resin processing to obtain the various articles is also performed in air.

[0013] It has now been unexpectedly found that it is possible to obtain polyester resins with a low formaldehyde content (less than 150 ppb) and with a low formaldehyde generation rate, lower than 15 ppb/minute at 290°C, by addition to a polyester resin of antioxidants capable of removing free radicals (free-radical scavengers).

[0014] When the resin is regraded in the solid state, in addition to containing the antioxidant which acts as free-radical scavenger, said resin must be heated in an inert-gas atmosphere.

[0015] The behavior of the polyester resins added with antioxidants acting as free-radical scavengers is unusual with respect to the formation of formaldehyde if one considers that by working under the same conditions it is not possible to limit neither the acetaldehyde content nor its generation rate.

[0016] Free-radical scavenging antioxidants are added to the polyester resin during the preparation of said resin in the esterification or transesterification step or in the subsequent step of polycondensation in the melt.

[0017] Said antioxidants can also be added to the resin in the melt by mixing in an extruder before forming the articles or before subjecting the resin to the SSP step.

[0018] The treatment of the resin in the melted state can be performed in an inert-gas atmosphere (nitrogen), for example by passing a stream of inert gas flow in the environment in which preparation of the articles occurs. It is also possible to work in air, but less satisfactory results are obtained.

[0019] The antioxidants are used in an amount from 0.1 to 1% by weight on the resin.

[0020] It has been found, and it is a further aspect of the invention, that it is possible to avoid coloring problems due to the presence of the antioxidant, particularly if the antioxidant is of the phosphite class, which occur when the resin is subjected to SSP at a temperature above 180°C, if the resin, which contains the antioxidant, is added with a dianhydride of a tetracarboxylic aromatic acid and is subjected to SSP at temperatures lower than 180°C.

[0021] Pyromellitic dianhydride is the preferred compound.

[0022] Copolyethylene terephthalate resins containing up to 15% isophthalic acid, with the addition of less than 0.2% pyromellitic dianhydride by weight, are particularly suitable for this purpose.

[0023] When working at temperatures lower than 180°C, for example between 170° and 180°C, the kinetics of the polycondensation reaction is still sufficiently high. The resulting polymer is colorless (yellow index of less than 4 measured according to ASTM D 542-50).

[0024] As already indicated, the antioxidants capable of acting as free-radical scavengers are usable in the process according to the present invention.

[0025] This category includes phosphites having the formula P(OR¹)₃, where the radicals R¹ are mutually identical or different and are alkyl or aryl groups, such as for example alkyl radicals with 8 or more carbon atoms, such as isooctyl,

isodecyl, octadecyl, or aryl radicals such as phenyl and phenyl substituted with one or more alkyl groups, such as t.butyl-phenyl, di-t.butyl-phenyl, n-nonyl-phenyl and the like.

[0026] It is also possible to use phosphates having the formula OP(OR¹)₃, where R¹ has the above mentioned meaning.

[0027] Triarylphosphates are preferred.

[0028] Diphosphite compounds such as the compound marketed by General Electric Specialty Chemicals under the name "Ultranox 626" or "Ultranox 627" are particularly suitable thanks to the low yellow index of resins stabilized with these antioxidants.

[0029] Examples of triphosphites are triphenylphosphite and trioctadecylphosphite.

[0030] Other antioxidants which can be used are phenolic antioxidants, such as hindered phenolic compounds, for example the compound "Irganox 1010" by Ciba-Geigy, constituted by tetrakis (methylene-3.5-di-t.butyl-4-hydroxy idrocinnamate).

[0031] The polyester resins that can be used in the process according to the invention are constituted by the product of the polycondensation of an aromatic dicarboxylic acid, such as for example terephthalic acid, with a diol with 2-12 carbon atoms, such as for example ethylene glycol, 1,4-butane diol and 1,4 cyclohexane dimethylol.

[0032] The preferred resins are constituted by polyethylene terephthalate, polybutylene terephthalate, and terephthalate copolymers in which up to 15% in mols of the units derived from the terephthalic acid are substituted with units derived from isophthalic acid or naphthalene dicarboxylic acids.

[0033] Resins with a low formaldehyde content and low formaldehyde generation rate have an intrinsic viscosity of more than 0.4 dl/g. Those with an intrinsic viscosity of more than 0.7 dl/g obtained by regrading in the solid state are particularly suitable for preparing containers for carbonated beverages, films, foamed materials for use, for example, for preparing trays for food and for any other application which requires a low formaldehyde content and a low formaldehyde generation rate.

[0034] The following examples are given by way of nonlimitative illustration of the invention.

Analytical methods

[0035] Formaldehyde determination is performed by HPLC (High-Pressure Liquid Chromatography), using dinitrophenylhydrazine (DNPH) as detection agent. The sample, placed in a flask adapted for gas flushing and equipped with a filtering disk, is desorbed at 150°C for 6 hours in a 10 ml/min nitrogen flow.

[0036] The compounds with carbonyl groups removed in this manner are fixed on highly pure silica coated with DNDH (a product marketed by Supelco, cat. no. 2-1026), where they are converted into the hydrazone derivatives.

[0037] The hydrazone derivatives are eluted by a cartridge with acetonitrile and water and analyzed by HPLC according to the following procedure.

Flow rate

35

40

45

1 ml/min

Injected volume

20 ml

Detection

UV, 360 nm

Column characteristic

C 18 4.6 mm x 250 mm I.D.

Intrinsic viscosity is determined on a solution of 0.5 g of polymer in 100 ml of a mixture 60/40 by weight of phenol/tet-rachloroethane at 25°C.

EXAMPLES

1. Production of dumbbells

[0038] The resin used is a polyethylene terephthalate (TPA) obtained from terephthalic acid by using an antimony catalyst (Sb triacetate) and a TPA copolymer with 5% isophthalic acid (IPA). The resin, dried beforehand, is obtained with the addition, in a screw extruder (Haake), of the antioxidants and of the pyromellitic anhydride (when used). The initial viscosity of the polymer was 0.579 dl/g; final viscosity ranges between 0.572 and 0.579 dl/g (see table).

[0039] The polymer thus produced is subjected to solid-state polycondensation in a 5.5-1 bench reactor, heated with an oil jacket and a stream of hot nitrogen, capable of maintaining the chips at a constant temperature of 196°C (examples 1, 2, 3) and 171°C (example 4). At the end of the polycondensation, the polymers had an intrinsic viscosity between 0.746 and 0.790 dl/g (see table).

[0040] These polymers, after being dried at 140°C, were used to prepare dumbbell specimens in a nitrogen-air atmosphere, by using a machine with an injection press (Arburg) at 300°C.

[0041] The results are summarized in Table 1.

2. Production of preforms

[0042] Starting from the same commercial polymer previously described, preforms were produced both with the polymers to which the antioxidants were added, as previously mentioned, and by adding the antioxidants directly in the mixer between the dryer and the extruder of the mold for the preforms (examples 5-6). The polymer had an initial viscosity of 0.762 dl/g and had been dried to 140°C.

[0043] The results are listed in Table 1.

[0044] Table 1 also reports the data related to the polymer without the addition of the antioxidants.

[0045] Results similar to those listed in the table are achieved by using resins prepared by using titanium catalysts (titanium dioxide-silica: AK20 C94).

[0046] The disclosures in Italian Patent Application No. MI98A000537 from which this application claims priority are incorporated herein by reference.

15

20

25

30

35

40

45

50

Table 1

5

10

15

20

25

30

35

40

45

50

55

	Comp. Ex.	Example 1	Example 2	Example 3	Example 4	Example 5	Example 6
Polymers TPA	×	×	×	×	×	×	\$ x
Copolymer					*		
TPA + 4.5% IPA			***************************************		:		
Ultranox 627 A		0.5%			0.5%	0.5%	
Etanox 330			0.5%				
ADK PEP 36				0.5%			0.5%
PMDA					0.2%		
Amorphous chips	×	×	×	×			
IV dl/gr	0.578	0.552	0.538	0.579			
Formaldehyde free ppt							
generation 1 min.	42	p4	5	2			
5 min.	371	28	56	30			
10 min.	2128	116	71	119			
Regraded chips	×	×	(1) x	×	x (1)	×	×
IV dl/gr	0.746	0.751	0.757	0.790	0.790	0.790	0.790
yde fre		*					
generation 1 min.	64	n.d.	∞	n.d.	n.d.		
5 min.	476	25	37	19	n.d.	,	
10 min.	3426	58	64	41	41		
Dumbbell	×	×	x (2)	×	x (2)		
Formaldehyde free ppt							
air	1740	38	89	47	e		
nitrogen	350	34	20	37	n.d.		
Preforms .	×	×	x (2)	X	x (2)	×	×
Formaldehyde free ppt							
air	962	20	40	22	n.d.	75	41
nitrogen	541	18	25	16	n.d.	46	36

: Bis (2,4-di-t-butylphenil)pentaerythritol disphosphite ULTRANOX 627 A

: 1,3,5,-Trimethyl-2,4,6-tris(3,5-di-tert-butyl-4-hydroxybenzyl)benzene : Bis (2,6-di-tert-butyl-4-methylphenyl)pentaerythritol-di-phosphite

ETANOX 330 ADK STAB PEP-36 (1) (2)

: product of good colour

: transparent product of good colour

Claims

5

10

15

20

- Aromatic polyester resins containing a free-radical scavenger antioxidant, characterized by a residual formaldehyde content of less than 150 ppb and by a formaldehyde generation rate at 290°C of less than 15 ppb/minute.
- 2. Resins according to claim 1, having an intrinsic viscosity of more than 0.7 dl/g.
- 3. Resins according to claims 1 and 2, chosen from the group consisting of polyethylene terephthalate, polybutylene terephthalate, copolyethylene terephthalate, containing up to 15% of isophthalic acid units.
- Resins according to claims 1 and 2, constituted by copolyethylene terephthalate containing up to 15% isophthalic
 acid added with less than 0.2% pyromellitic dianhydride.
- 5. Articles obtained from the resins of the preceding claims 1 to 4.
- 6. Preforms for beverage bottles and related bottles obtained from the resins of claims 2, 3 or 4.
- 7. A method for preparing the resin according to claims 1 to 4, wherein an antioxidant which has a free-radical scavenging action is added to the resin in molten state and the resin is then subjected to a solid-state polycondensation reaction in an inert-gas atmosphere.
- 8. A method according to claim 7, wherein a dianhydride of an aromatic tetracarboxylic acid is added to the resin and the solid-state polycondensation reaction is performed at temperatures from 170° to 180°C.
- 25 9. Colorless resins obtainable according to the method of claim 8.
 - 10. Resins and articles according to the preceding claims 1 to 6 and 9, containing an antioxidant chosen among the classes formed of phosphites and/or organic phosphates and hindered phenols.
- 30 11. Resins according to claim 10, wherein the antioxidant is added to the resin in an amount from 0.1 to 1% by weight.

35

40

45

50



Europäisches Patentamt

European Patent Office

Office européen des brevets



(11) EP 0 943 649 A3

(12)

EUROPEAN PATENT APPLICATION

(88) Date of publication A3: 07.02.2001 Bulletin 2001/06

(51) Int. Cl.⁷: **C08K 5/00**, C08K 5/13, C08K 5/52, C08L 67/02

(43) Date of publication A2: 22.09.1999 Bulletin 1999/38

(21) Application number: 99104011.4

(22) Date of filing: 12.03.1999

(84) Designated Contracting States:

AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU

MC NL PT SE

Designated Extension States:

AL LT LV MK RO SI

(30) Priority: 17.03.1998 IT MI980537

(71) Applicant: SINCO RICERCHE S.p.A. 28048 Verbania Pallanza (Verbania) (IT)

(72) Inventors:

Al Ghatta, Hussein
 03014 Fiuggi (Frosinone) (IT)

• Pinto, Antonio 00042 Anzio (Roma) (IT)

(74) Representative:
 Modiano, Guido, Dr.-Ing. et al
 Modiano & Associati SpA
 Via Meravigli, 16
 20123 Milano (IT)

(54) Stabilized polyester resins having reduced formaldehyde content

(57) Polyester resins with a low formaldehyde content, to which an antioxidant acting as free-radical scavenger is added, containing less than 150 ppb of formaldehyde and having a formaldehyde generation rate of less than 15 ppb/min at 290°C.



EUROPEAN SEARCH REPORT

Application Number

Category	Citation of document with in of relevant pass	ndication, where appropriate, ages	Refevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.6) C08K5/00 C08K5/13 C08K5/52 C08L67/02	
Y	DATABASE CHEMABS [CHEMICAL ABSTRACTS OHIO, US; retrieved from STN Database accession HCAPLUS XP002155023 * abstract * & JP 09 012698 A (MIND.) 14 January 19	SERVICE, COLUMBUS, no. 126:187208/DN,	1-11		
Y	WO 93 08226 A (M & 29 April 1993 (1993 * claims 1,8,10; ex * page 7, paragraph	-04-29) amples 1-3: table 1 *	1-11		
Υ	DATABASE WPJ Section Ch, Week 19 Derwent Publication Class A23, AN 1996- XP002155024	s Ltd., London, GB:	1-7		
-	& JP 08 027258 A (M	ITSUI PETROCHEM IND CO		TECHNICAL FIELDS SEARCHED (Int.CL6)	
	LTD), 30 January 19 * abstract *	96 (1996-01-30)		C08G C08K	
Υ }	DD 240 309 A (ADW D 29 October 1986 (19 * the whole documen	86-10-29)	1-7	-	
A	AND STABILITY,	POLYMER DEGRADATION s 393-397, XP000926098	3 1		
A WO 97 44376 A (WEL 27 November 1997 (* abstract; examp)		997-11-27)	1-7,10		
	•	-/			
<u>+</u>	The present search report has b	peen drawn up for all claims			
·	Place of search	Date of completion of the search		Examiner	
	BERLIN	8 December 2000) Kı	rische, D	
X : parti Y : parti docu	ATEGORY OF CITED DOCUMENTS cularly relevant if taken alone cularly relevant if combined with anoth ment of the same category nological background	L : document cite	document, but pu date ed in the apolication	blished on, or on os	



EUROPEAN SEARCH REPORT

Application Number EP 99 10 4011

Category	Citation of document with indica of relevant passages		Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.6)
D,A	EP 0 041 035 A (RHONE- 2 December 1981 (1981- * claims 1,2; example	-POULENC IND.) -12-02)	1-6	
				·
ļ				
				TECHNICAL FIELDS SEARCHED (Int.Cl.6)
	·			
·				
	The present search report has been	drawn up for all claims	-	
	Place of search BERLIN	Base of completion of the search 8 December 2000	Kri	Sche, D
X : parti Y : parti docu	ATEGORY OF CITED DOCUMENTS icularly relevant if taken alone cularly relevant if combined with another unent of the same category nological background	T : theory or principl E : earlier patent do after the filing da D : document cited t L : document cited t	le underlying the incument, but publiste te in the application	

ANNEX TO THE EUROPEAN SEARCH REPORT ON EUROPEAN PATENT APPLICATION NO.

EP 99 10 4011

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

08-12-2000

cited in search rep	t ort	Publication date	Patent family member(s)	Publication date
JP 9012698	Α	14-01-1997	NONE	
WO 9308226	Α	29-04-1993	IT 1251953 B	27-05-199
		,	AU 664388 B	16-11-199
			AU 2758392 A	21-05-199
			CA 2096640 A	19-04-199
			EP 0563354 A	06-10-199
			JP 2790917 B	27-08-199
			JP 6503606 T	21-04-199
			US 5376734 A	27-12-199
JP 8027258	Α	30-01-1996	NONE	
DD 240309	Α	29-10-1986	NONE	
WO 9744376	Α	27-11-1997	US 5898058 A	27-04-199
			AU 3010197 A	09-12-199
			BR 9709324 A	04-01-200
			CN 1219181 A	09-06-199
			EP 0900247 A	10-03-199
			JP 2000510180 T	08-08-200
EP 0041035	Α	02-12-1981	FR 2482971 A	27-11-198
			AT 10375 T	15-12-198
			AU 540702 B	29-11-198
			AU 7065681 A	26-11-198
			BR 8103108 A	09-02-198
			CA 1164144 A	20-03-198
			DE 3167283 D	03-01-198
	•		ES 502313 D	01-10-198
			ES 8300121 A JP 1021167 B	01-01-198
				20-04-198
			JP 1545189 C JP 57016024 A	15-02-199
			KR 8400468 B	27-01-198 09-04-198
			MX 166409 B	07-01-199
			PT 73056 A.B	01-06-198
			US 4340721 A	20-07-19
				6
		· W.		